Short Communication

LOGISTIC HEATING PROGRAM IN THERMAL ANALYSIS AND NON-ISOTHERMAL KINETICS

V. M. Gorbachev

INSTITUTE OF INORGANIC CHEMISTRY, SIBERIAN DEPARTMENT OF THE ACADEMY OF SCIENCES OF THE SOVIET UNION, NOVOSIBIRSK, U. S. S. R.

(Received December 3, 1982)

Use of the logistic rule is proposed for programmed heating in thermal analysis. An algorithm is presented for solving the temperature integral of non-isothermal kinetics.

In thermoanalytical practice a sharp difference is made between the study of transformations under isothermal and non-isothermal conditions. Formally, though, the isothermal process may be considered the boundary case of the non-isothermal process with the condition $\frac{dT}{dt} \rightarrow 0$. In the general case, the technique of isothermal investigations consists of two successive stages: the non-programmed (or in some cases the linearly programmed) heating of the sample to a selected temperature, and the subsequent keeping of the sample at this temperature until the transformation is ended. The idea to replace the non-isothermal and isothermal stages by one program combined analytically is an obvious one. The logistic rule of heating appears suitable for such a program. It is expressed mathematically by the known Perl equation (1):

$$\frac{\mathrm{d}T}{\mathrm{d}t} = cT(T_0 - T) \tag{1a}$$

or

$$T = \frac{T_0}{1 + \lambda \exp\left(-cT_0t\right)} \tag{1b}$$

where c and λ are constants and T_0 is the upper boundary of temperature rise. For logistic temperature programming the kinetic equation

$$\frac{d(\alpha)}{dt} = A \exp{-\frac{E}{RT}} f(\alpha)$$
(2)

J. Thermal Anal. 28, 1983

can be written in the following form:

$$g(\alpha) = \int_{0}^{\alpha} \frac{d(\alpha)}{f(\alpha)} = \frac{A}{c} \int_{0}^{T} \frac{\exp\left(-\frac{E}{RT}\right)}{T(T_0 - T)} dT$$
(3)

(the symbols being those generally applied for non-isothermal kinetics).

The following task is to find an integration procedure for the kinetic equation (3). For this purpose, starting from the train of thought of Frank-Kamenetsky [2, 3], it is necessary to transform the expression $(T_0 - T)$ into an exponential function, that is $(T_0 - T) = a \exp \frac{b}{RT}$, or $\ln (T_0 - T) = \ln a + \frac{b}{RT}$, where a and b are constants. Utilizing the Frank-Kamenetsky transformation, the kinetic equation [3] will assume the form

$$g(\alpha) = \frac{A}{ac} \int_{0}^{t} T^{-1} \exp\left(-\frac{E+b}{RT} dT\right)$$
(4)

The solution of this temperature integral has already been reported in [4] for the condition $E \gg RT$; its general form is

$$\int_{0}^{T} T^{Z} \exp\left(-\frac{E}{RT}\right) dT = \frac{RT^{z+2}}{E + (z+2)RT} e^{-E/RT}$$

Finally, after the necessary mathematical transformations, one obtains:

$$g(\alpha) = \frac{A}{ac} \frac{RT}{(E+b) + RT} \exp\left(-\frac{E+b}{RT}\right)$$

or

$$g(\alpha) = \frac{A}{c} \frac{RT \exp \left(-\frac{L}{RT}\right)}{(T_0 - T)(E + b + RT)}$$
(5)

Consequently, by performing the thermoanalytical experiment according to the logistic heating program, one can determine the kinetic value of the activation energy E in the interval of the exponential rise of temperature in this program.

Besides the Perl curve, another logistic curve, that of Gompertz [1], is also known:

$$T = T_0 \exp\left(-\lambda e^{-ct}\right)$$

However, its utilization for temperature programming does not provide any particular advantage, while the difficulty of solving the temperature integral increases substantially.

J. Thermal Anal. 28, 1983

References

- 1 J. Martino, Technological forecasting. Progress, Moscow, 1977, p. 124, 126 (in Russian).
- 2 D. A. Frank-Kamenetsky, Diffusion and heat transfer in chemical kinetics. Nauka, Moscow, 1967 (in Russian).
- 3 Ya. B. Zelkovits, B. I. Barenblatt, V. B. Librovich and G. M. Makhviladze, The mathematical theory of burning and explosion. Nauka, Moscow, 1980, p. 48 (in Russian).
- 4 V. M. Gorbachev, J. Thermal. Anal., 10 (1976) 447.